O(20) carbonyl oxygen would readily add to the reactive C(1) atom of phorbol in a 'Micheal-type' addition reaction. Such a mechanism has been suggested for the antitumor action of α -methylene lactones¹⁹.

As additional support for the relationship of cortisol and phorbol, we have determined that phorbol (1% cross reaction) and PMA (6% cross reaction) are 'recognized' as cortisol at ca. $10^{-8} M$ in a radioimmunoassay 20,21 .

The molecular similarity of 9α -fluoro-cortisol 1 and phorbol 2

	Cortisol atom ¹⁴	Phorbol atom 15	Delta (Å)
Least squares fit			
of these atoms	O(17)	O(4)	0.322
	O(21)	O(9)	0.348
	O(20)	C(1)	0.577
	O(11)	O(20)	0.263
Unmatched atoms	C(13)	C(5)	0,533
	C(17)	C(4)	0.472
	C(20)	C(10)	0.346
	C(16)	C(3)	0.498
	C(12)	C(6)	1.546
	C(21)	C(9)	0.239
	C(11)	C(20)	2.309

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- ¹⁸ G. CALCUTT, Br. J. Cancer 15, 390 and 855 (1961).
- ¹⁹ S. M. Kupchan, D. C. Fessler, M. A. Eakin and T. J. Giacobbe, Science 168, 376 (1970).
- ²⁰ Gammacoat¹²⁵I Cortisol Radioimmunoassay (Clinical Assays, Inc., 237 Binney Street, Cambridge, Massachusetts, USA).
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Dehydroacetic Acid in Anthers of Solandra nitida (Solanaceae)

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Summary. The chloroform extract of anthers of Solandra nitida contains 3-acethyl-6-methyl-1,2-pyran-2,4(3H)dione (dehydroacetic acid) as one of the main products.

The size of certain tropical flowers allow the isolating of chemical compounds from the different parts of the flower. We have collected and extracted a quantity of anthers of Solandra nitida (solanaceae) called 'golden cup'. The first compound that was found was saccharose from the methanolic extract 1 besides hexaeicosanol, 8-pentaeicosanol and nonaeicosane 2. Now we report the presence of dehydroacetic acid.

1 kg of anthers were extracted with hexane and then with chloroform. By TLC the chloroform extract showed 14 spots in toluene-ethyl formiate-formic acid (5/5/1). By column chromatography in the same eluent, we got 13 different fractions. The 8th fraction was crystallized from toluene-AcOEt (5/5) and gave yellow needles, m. p. $100-104\,^{\circ}\text{C}$, recrystallization in AcOEt–MeOH (1/5) gave amber needles m. p. $109-110\,^{\circ}\text{C}$. The MS showed that the compound has 8 carbon atoms $(M+1=8.5\%)^3$, molecular weight 168 and formula $C_8H_8O_4$. The IR-spectrum showed the following bands; $3065\,\text{cm}^{-1}$ (w) CH=C, $1720\,$ (s) C=O lactone and diketone, $1450\,$ (m) C=CH, $1375\,$ (m) CH₃, $1255\,$ CH₃-C=O O=C -O-C, $1000\,$ (s) C=CH out of plane, $860\,$ (m)-O-C-CH₃, $780\,$ (m) and $712\,$ (w) C=CH st. UV-spectrum showed 2 bands at λ 223 nm (log E = 3.9) and

 λ_{max} at 308 nm (log E = 1.2). The NMR-spectrum showed 3 signals at δ in ppm 2.33 (s) (3H) CH₃–CO, 2.66 (s) (4H) CH₃C=C and $\stackrel{1}{-}$ C-C=O, 5,95 (m) (1 H) vinylic proton. All O- H

these data suggest that the compound is 2,6-dioxo-2-methyl-5-acethyl-dihydropyrane (dehydroacetic acid). The mass spectrum confirms this structure because we observe the following fragments; m/e 168, 153, 125, 110, 69, 56, 43 (base peak).

To our knowledge it is the first time that dehydroacetic acid is found in nature, but it is well known as the dimerization product of ethyl acetoacetate. The fact that it was found in the masculine sexual organ of the flower (anthers) suggests a metabolic activity related to the reproduction, e.g. a way for anabolism different to the well-known one through mevalonic acid.

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